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INSTITUTE OF AERONAUTICAL ENGINEERING

(Autonomous)

B.Tech III Semester End Examinations (Regular) - December, 2017

Regulation: IARE – R16

THERMODYNAMICS
(Mechanical Engineering)

Time: 3 Hours

Max Marks: 70

Answer ONE Question from each Unit

All Questions Carry Equal Marks

All parts of the question must be answered in one place only

UNIT – I

1. (a) Derive an expression for the non-flow displacement work done during adiabatic process given by $PV^\gamma = C$, where $\gamma = C_p/C_v$ [7M]
- (b) A closed system undergoes two processes one after the other-constant pressure process at pressure of 5 bar from initial volume of 0.03 m^3 to 0.09 m^3 it is followed by polytropic expansion. $PV^{1.3}=C$ from 0.09 m^3 volume to 0.02 m^3 final volume find [7M]
 - i. Final pressure after expansion
 - ii. Work done during each process and net work done
2. (a) Explain Joules experiment with reference to PMM1. [6M]
- (b) Air flows steadily at the rate of 0.4 kg/s through an air compressor, entering at 6 m/s with a pressure of 1 bar and a specific volume of $0.85 \text{ m}^3/\text{kg}$, and leaving at 4.5 m/s with a pressure of 6.9 bar and a specific volume of $0.16 \text{ m}^3/\text{kg}$. The internal energy of the air leaving is 88 kJ/kg greater than that of the air entering. Cooling water in a jacket surrounding the cylinder absorbs heat from the air at the rate of 59 W . Calculate the power required to drive the compressor and the inlet and outlet cross-sectional areas. [8M]

UNIT – II

3. (a) Distinguish between heat engine and heat pump. [5M]
- (b) Two reversible heat engines A and B are arranged in series, A rejecting heat directly to B. Engine A receives 200 kJ at a temperature of 421°C from a hot source, while engine B is in communication with a cold sink at a temperature of 4.4°C . If the work output of A is twice that of B, find [9M]
 - i. The intermediate temperature between A and B
 - ii. The efficiency of each engine
 - iii. The heat rejected to the cold sink
4. (a) What is Carnot's cycle? Explain in detail with relevant sketches and processes. [6M]
- (b) Two kg of water at 80°C are mixed adiabatically with 3kg of water at 30°C in a constant pressure process of 1 atmosphere. Find the increase in entropy of the total mass of water due to mixing process. [8M]

UNIT – III

5. (a) Draw the phase equilibrium diagram for pure substance on h-s plot with relevant constant property lines and why the isobars lines are diverges from one another? [7M]
- (b) A vessel of volume 0.04 m^3 contains a mixture of saturated water and saturated steam at a temperature of 250°C . The mass of the liquid present is 9 kg. Find pressure, the specific volume, the enthalpy, the entropy. [7M]
6. (a) Derive an expression for the Vander Walls constants 'a' and 'b' in terms of critical properties. [7M]
- (b) 1 kg of CO_2 has a volume of 0.86 m^3 at 120°C . Compute the pressure using
- Ideal gas equation [7M]
 - Vander Wall's Equation
- Take Vander Wall's constant for CO_2 , $a = 365.6 \text{ kNm}^4/(\text{kg mole})^2$ and $b = 0.0423 \text{ m}^3/(\text{kg mole})$

UNIT – IV

7. (a) Define mole fraction and mass fraction. [4M]
- (b) 0.5 kg of helium and 0.5 kg of nitrogen are mixed at 20°C and at a total pressure of 100 kPa. Find [10M]
- the mole fraction of each constituent
 - Equivalent molecular weight of the mixture
 - the equivalent gas constant of mixture
 - the partial pressures and volumes
 - C_p and C_v of the mixture.
8. (a) Explain [7M]
- Dew point temperature
 - Degree of saturation
 - Adiabatic saturation process
- (b) An air water vapour mixture enters an adiabatic saturator at 30°C and leaves at 20°C , which is the adiabatic saturation temperature. The pressure remains constant at 100kPa. Determine the relative humidity and the humidity ratio of the inlet mixture. [7M]

UNIT – V

9. (a) For the same compression ratio and heat rejection, which cycle is most efficient; Otto, diesel or dual? Explain with P-V and T-S diagrams. [7M]
- (b) An ideal diesel cycle with air as the working fluid has a compression ratio of 18 and a cut-off ratio of 2. At the beginning of compression, the air is at 100kPa, 27°C and 1917 cm^3 . Determine
- the pressure and temperature of air at each point [7M]
 - the net work and thermal efficiency

10. (a) Derive air standard efficiency of Brayton cycle with P-V and T-S diagram [7M]
- (b) An air refrigeration open system operating between 1 mpa and 100 kpa is required to produce cooling effect of 2000 kJ/min. the temperature of air leaving the cold chamber is -5°C and at leaving the cooler is 30°C . Neglecting losses and clearance in the compressor and expander determine [7M]
- i) Mass of air circulated per hour
 - ii) Compressor work, expander work, cycle work
 - iii) Coefficient of performance and power required to run the machine

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