

**MECHANICAL ENGINEERING****ATTAINMENT OF COURSE OUTCOME - ACTION TAKEN REPORT**

|                      |                              |               |                               |
|----------------------|------------------------------|---------------|-------------------------------|
| Name of the faculty: | <b>Mr. M PRAVEEN</b>         | Department:   | <b>Mechanical Engineering</b> |
| Regulation:          | <b>IARE - R18</b>            | Batch:        | <b>2018-2022</b>              |
| Course Name:         | <b>Engineering Chemistry</b> | Course Code:  | <b>AHSB03</b>                 |
| Semester:            | <b>II</b>                    | Target Value: | <b>60% (1.8)</b>              |

**Attainment of COs:**

|     | Course Outcome  | Direct attainment | Indirect attainment | Overall attainment | Observation  |
|-----|---|-------------------|---------------------|--------------------|--------------|
| CO1 | Explain the operation of electrochemical systems for the production of electric energy, i.e. batteries.   | 0.30              | 2.30                | 0.7                | Not Attained |
| CO2 | Utilize electrochemical cell parameters, electrochemical active surface area, current and over potential under given condition for calculating the electromotive force and electrode potential. | 0.00              | 2.20                | 0.4                | Not Attained |
| CO3 | Illustrate the chemical and electrochemical corrosion in metals by influencing the nature of environment.   | 1.00              | 2.20                | 1.2                | Not Attained |
| CO4 | Make use of the basic electrochemical knowledge of corrosion processes for protection of different metals from corrosion.   | 0.00              | 2.30                | 0.5                | Not Attained |
| CO5 | Identify the hardness of water for finding the hardness causing salts in water.   | 0.00              | 2.30                | 0.5                | Not Attained |
| CO6 | Demonstrate different treatment methods for producing soft water from saline or brackish sources.   | 1.00              | 2.20                | 1.2                | Not Attained |

**Action Taken:**

CO1: More examples need to be given for the operation of electrochemical systems.

CO2: More assignments may be given for calculating the electromotive force and electrode potential.

CO3: More applications may be given for the corrosion process.

CO4:  
More examples may be given on the protection of different metals from corrosion.

CO5: More problems to be solved for finding the hardness causing salts in water.

CO6: More assignments may be given on different treatment methods for producing soft water from saline or brackish sources.

  
Course Coordinator



  
Head of the Department

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