

INSTITUTE OF AERONAUTICAL ENGINEERING

(Autonomous) Dundigal, Hyderabad -500 043

MECHANICAL ENGINEERING

TUTORIAL QUESTION BANK

Course Name	METALLURGY AND MATERIAL SCIENCE
Course Code	AME005
Class	III Semester
Branch	Mechanical Engineering
Year	2018-2019
Course Coordinator	Mr. M Prashanth Reddy, Assistant Professor
Course Faculty	Mr. M.V. Aditya Nag, Assistant Professor
	Mr. M Prashanth Reddy, Assistant Professor

COURSE OBJECTIVES:

The course should enable the students:

I	Understanding of metallurgical engineering concepts and properties.			
II	Analyze microstructures of metals and alloys and relationship to heat treatment.			
III	Compare the properties of ceramics, glasses, composites and polymers for industrial applications.			

COURSE LEARNING OUTCOMES:

Students, who complete the course, will have demonstrated the ability to do the following:

CAME005.02 Explain the necessity of alloying, types of solid solution and intermediate alloy phases. CAME005.03 Explain the concept of phase and phase diagram and understand the basic terminologies associated with metallurgy. CAME005.04 Construction of phase diagrams and identification of different phases and invariant reaction. CAME005.05 Understand and suggest the heat treatment processes and types, and significance of mechanica and metallurgical properties with respect to microstructures. CAME005.06 Explain the concept of Hardenability and demonstrate the test used to find the Hardenability of steels. CAME005.07 Analyze the microstructure of metallic materials using phase diagram and modify the microstructure and properties using different heat treatment processes. CAME005.08 Define and differentiate engineering materials on the basis of structure and properties of engineering applications. CAME005.09 Explain features, classification, and application of materials like polymers like thermosetting thermoplastics. CAME005.10 Explain features, classification, and application of materials like ceramics. CAME005.11 Explain features, classification, and application of materials like ceramics, composites and polymers. CAME005.12 Differentiate the properties and application of various materials like ceramics, composites and polymers. CAME005.13 Make the students conversant with ISO and IS standards of the material composition and mechanical properties. CAME005.14 Design and develop materials for high temperature applications and understand mechanical properties at elevated temperatures.		
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CAME005.16 Design the materials for strength to weight ratio applications.	CAME005.16	Design the materials for strength to weight ratio applications.

	UNIT – I		
	STRUCTURE OF METALS		
	Part - A (Short Answer Questions)		
S No	QUESTION	Blooms	Course
5110	QUESTION	Taxonomy	Learning
		Level	Outcomes
1	Define Crystallography and Explain Unit Cell	Understand	CAME005.01
2	Define unit cell and Space lattice with a neat diagram.	Understand	
3	Define space lattice with a neat diagram.	Remember	CAME005.01
4	Define alloy. Explain necessity of alloy elements with few examples.	Remember	CAME005.02
5	What is grain? Explain the concept of grain boundary.	Remember	CAME005.01
6	Explain miller indices. Draw the miller indices plane for (100).	Understand	CAME005.01
7	Explain miller indices. Draw the miller indices plane for (101).	Understand	CAME005.01
8	Define miller indices. Draw the miller indices plane for (110).	Understand	CAME005.01
9	Explain miller indices. Draw the miller indices plane for (111).	Understand	CAME005.01
10	Define miller indices. Draw the miller indices plane for $(\bar{1}10)$.	Understand	CAME005.01
11	What are imperfections in crystals? Explain Point defect with a	Remember	CAME005.01
	diagram?		
12	What is defect? Explain Line defect with a neat diagram?	Remember	CAME005.01
13	Classify the line defects. Define Surface defect?	Remember	CAME005.01
14	What is Volume defect? Explain volume defect with neat sketch.	Remember	CAME005.01
15	Define Dislocation?	Remember	CAME005.01
16	Define the term Alloy. Name the alloys with their properties.	Understand	CAME005.02
17	What is Phase? Explain with a phase diagram.	Understand	CAME005.03
18	Define the term Solid solution.	Remember	CAME005.03
19	Define the term Intermediate phase.	Understand	CAME005.03
20	Explain the term Electron compound.	Understand	CAME005.03
	Part - B (Long Answer Questions)		
1	Define grain and grain boundary? Explain about are the characteristics	Understand	CAME005.01
	of grain boundary?		
2	Define grain size? Explain about methods for determining grain size?	Remember	CAME005.01
3	Explain effect of grain size on properties.	Remember	CAME005.01
4	State and explain Hume Rothery rules for the formation of solid	Understand	CAME005.03
	solutions.		
5	Explain about intermediate phases? Explain about are the various	Understand	CAME005.03
	types of intermediate phases?	TT1. / 1	CAMEOUTOS
6	Explain about crystalline material? Distinguish between single crystal	∪nderstand	CAME005.03
7	material and polycrystalline material? Define the terms (i) Space lattice (ii) Unit cell (iii) Solid solution and	Remember	CAME005.01
'	monotectic solution.	Kemember	CAMEUUS.UI
8	Explain the procedure to find out the miller indices with an example.	Understand	CAME005.01
U	England the procedure to find out the finner findices with all example.	- Haci stand	51 II.IL 005.01

9	Draw the miller indices for (i) (100) (ii) (110) (iii) (111).	Remember	CAME005.01	
10	List out and draw various bravias lattice structures and their primitives.	Remember	CAME005.01	
11	Write about relation between a, b, c and α , β , γ in cubic crystal system, tetragonal crystal system, orthorhombic crystal system and Hexagonal crystal system.	Remember	CAME005.01	
12	Define packing factor? Explain about is the packing factor for (i) Simple cubic crystal (ii)Body centred cubic crystal.	Understand	CAME005.01	
13	Explain about effective number of atoms? Calculate the effective number of atoms in SC structure, FCC structure, and BCC structure.	Understand	CAME005.01	
14	Explain about relation between lattice constant (a) and atomic radius(r) in SC structure, FCC structure, BCC structure, HCP structure?	Understand	CAME005.01	
15	Define coordination number. Explain about is the coordination number for BCC, FCC, HCP (Explain with proper procedure)?	Understand	CAME005.01	
16	Discuss about necessity of alloying?		CAME005.02	
17	Write briefly about Humme Rothery rules.	Remember	CAME005.02	
18	Explain about solid solutions?	Remember	CAME005.02	
19	Explain about types of solid solutions?	Remember	CAME005.02	
20	Describe about linear atomic density? Calculate the linear atomic density in [110] direction in the cooper crystal lattice in atoms per mm. copper is FCC and has a lattice constant of 0.351.		CAME005.01	
	Part - C (Problem Solving and Critical Thinking Question	ons)		
1	Compare simple cubic and body centred cubic crystal. Explain atomic packing factor.	Understand	CAME005.01	
2	Explain about coordination number?	Understand	CAME005.01	
3	Explain about atomic packing factor? Explain atomic packing factor for SC, BCC &FCC structures.	Understand	CAME005.01	
4	Explain covalent and ionic bonding. Compare covalent and ionic bond.	Remember	CAME005.01	
5	Describe about metallic bond? Compare covalent and metallic bond.	Remember	CAME005.01	
6	What is grain structure? Compare coarse and fine grain structure.	Remember	CAME005.01	
7	Explain defects. Compare point and line defects with diagrams.	Understand	CAME005.01	
8	Compare pure metals and alloys. Give the examples.	Understand	CAME005.02	
9	Compare substitutional and interstitial solid solutions. Give the examples.	Remember	CAME005.02	
10	Describe about melting range in alloys?	Remember	CAME005.02	
	UNIT - II			
PHASE DIAGRAMS				
	Part – A (Short Answer Questions)			
S No	QUESTION	Blooms Taxonomy	Course Learning Outcomes	
1	Explain the term Binary alloy.	Level Understand	CAME005.04	
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2	Define phase and Gibbs Rule.	Understand	CAME005.04
3	Define Gibbs Rule, levers rule.		CAME005.04
4	What is gibb's phase rule? Explain its importance.	Remember	CAME005.04
5	Define the term coring.		CAME005.04
6	What is Isomorphous system Define?		CAME005.04
7	Define Eutectic system.		CAME005.04
8	Define Partial eutectic system.		CAME005.04
9	Explain Peritectic system.		CAME005.04
10	Define Monotectic system.	Understand	CAME005.04
11	Explain about is melting range?		CAME005.04
12	Define Phase Diagram.		CAME005.04
13	Explain about are the types of substitutional solid solutions?	Remember	CAME005.04
14	Explain about is Substitutional solid solution?	Understand	CAME005.04
15	Define cooling curve.		CAME005.04
16	Define Eutectoid reaction.		CAME005.04
17	Write about Thermal Equilibrium Diagram?	Remember	CAME005.04
18	Draw the stages of structures from Solid to Liquid formation in binary	Understand	CAME005.04
10	system.	Understand	CAME003.04
19	Explain about intermediate phases using examples?	Remember	CAME005.04
20	Discuss about use of cooling curves?	Understand	CAME005.04
	Part - B (Long Answer Questions)		
1	Explain with the help of a diagram the cooling curve of pure metals.	Understand	CAME005.04
2	State and explain levers rule.	Remember	CAME005.04
3	Explain with the help of a diagram the cooling curve of alloys.	Understand	CAME005.04
4	State and explain Gibbs phase rule.	Understand	CAME005.04
5	Explain the construction of phase diagram.	Remember	CAME005.04
6	Explain about non equilibrium cooling.	Understand	CAME005.04
7	Write in brief about the binary phase diagram.	Remember	CAME005.04
8	Write about purposes of phase diagrams?	Understand	CAME005.04
9	Define the term isomorphism and polymorphism.	Understand	CAME005.04
10	Explain about the isomorphous system with a Ni-Cu diagram.	Understand	CAME005.04
11	Write a brief note about eutectic system.	Remember	CAME005.04
12	Explain the phase change in a eutectic system with an example.	Understand	CAME005.04
13	Write a short note on eutectoid system.	Understand	CAME005.04
14	Explain with an example the eutectoid system.	Remember	CAME005.04
15	Write about dendrites?	Understand	CAME005.04
16	Explain the formation of dendrites.	Understand	CAME005.04
17	Write about most common types of phase diagrams explain in brief?	Remember	CAME005.04
18	Draw and explain the Cd-Bi phase diagram.	Understand	CAME005.04
19	Explain about the levers rule and write its application.	Understand	CAME005.04
20	Draw and explain the cooling curves for pure metals.	Remember	CAME005.04
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	Part – C (Problem Solving and Critical Thinking)			
1	Compare eutectic and peritectic reaction.	Understand	CAME005.04	
2	Compare eutectic and eutectoid reaction.	Remember	CAME005.04	
3	Explain about peritectoid reaction?	Understand	CAME005.04	
4	Compare peritectoid and eutectoid reaction.	Understand	CAME005.04	
5	Explain about phase in detail?	Remember	CAME005.04	
6	Explain about Gibbs phase rule?	Understand	CAME005.04	
7	Compare phase and component.	Remember	CAME005.04	
8	Compare isomorphous and eutectic system.	Understand	CAME005.04	
9	Explain about coring?	Understand	CAME005.04	
10	Explain about intermediate phases?	Understand	CAME005.04	
	UNIT-III			
	ENGINEERING MATERIALS-I			
	Part - A (Short Answer Questions)			
S No	QUESTION	Blooms	Course	
		Taxonomy	Learning	
1	Define alloy. Explain different types of alloys.	level Understand	Outcomes CAME005.04	
2	Define phase and draw the phase diagram.	Understand		
3	Define Gibbs Rule.	Understand		
4	Explain levers rule.	Remember	CAME005.04	
5	Define ferrite and austenite.	Understand		
6	Explain Isomorphous system.	Understand		
7	Explain Eutectic system	Remember	CAME005.04	
8	Define Partial eutectic system.	Understand		
9	Write about Peritectic, eutectoid system.	Remember	CAME005.04	
10	Define Monotectic system.	Understand	CAME005.04	
11	Explain about Critical Temperature?	Remember	CAME005.04	
12	Define Phase Diagram.	Understand	CAME005.04	
13	Explain about are the types of substitutional solid solutions?	Understand	CAME005.04	
14	Explain about Substitutional solid solution?	Remember	CAME005.04	
15	Define cooling curve.	Understand	CAME005.04	
16	Define continuous cooling transformation curves.	Remember	CAME005.04	
17	Explain about is use of Thermal Equilibrium Diagram?	Remember	CAME005.04	
18	Draw the stages of structures from Solid to Liquid formation in binary	Remember	CAME005.04	
	system			
19	Describe about formation of intermediate phases in detail.	Remember	CAME005.04	
20	Explain about is the use of Jominy end Quenching Machine.	Understand	CAME005.07	
	Part – B (Long Answer Questions)			
1	Explain allotropic forms of iron and give lattice structure of each.	Understand	CAME005.04	

2	Define ferrite, pearlite, and austenite.	Understand	CAME005.04	
3	Explain about peritectic reaction and explain with diagram?	Understand	CAME005.04	
4	Explain about eutectic reaction in iron–carbide system and explain	Understand	CAME005.04	
	with neat diagram?			
5	Explain various phase reactions in iron-iron carbide system.	Understand	CAME005.04	
6	With the help of neat sketch explain eutectoid reaction.	Understand	CAME005.04	
7	Explain about hypo eutectoid steels and hyper eutectoid steels.	Understand	CAME005.04	
8	Define ferrite, austenite, cementite and martensite.	Understand	CAME005.04	
9	Explain various effects of sulphur, phosphorous additions in steel.	Understand	CAME005.04	
10	Explain in detail about isothermal transformation.	Understand	CAME005.04	
11	Draw time temperature transformation curves and identify products.	Understand	CAME005.04	
12	Draw continuous transformation curves and explain.	Remember	CAME005.04	
13	What is annealing heat treatment explain?	Understand	CAME005.05	
14	Explain normalizing heat treatment process.	Understand	CAME005.05	
15	Explain hardening and tempering heat treatment processes.	Remember	CAME005.05	
16	Compare stainless steels and tool steels.	Remember	CAME005.05	
17	Explain Hardenability and the method of testing using Jominy end	Understand	CAME005.06	
	quench.			
18	Classify stainless steels and mention their properties and applications.		CAME005.12	
19	Explain properties and applications of austenitic stainless steels.		CAME005.12	
20	Explain applications of Tool steels, HSLA steels and Maraging steels.	Understand	CAME005.13	
	Part – C (Problem Solving and Critical Thinking)			
1	Explain about is allotropy in iron?	Understand	CAME005.04	
2	Compare ferrite and austenite, ferrite and pearlite.	Understand	CAME005.04	
3	Draw Fe-C equilibrium diagram and label the temperatures,	Understand	CAME005.04	
	compositions and phases.		~	
4	Compare ferrite and cementite.		CAME005.04	
5	Differentiate between hypo and hyper eutectoid steels.	Understand	CAME005.04	
6	Compare stools and east irons as per earlier per percents as	Understand	CAME005.04	
7	Compare steels and cast irons as per carbon per percentage. Compare low and medium carbon steels.	Understand		
8	Explain following with neat diagram Peritectic transformations,	Understand		
0	Explain following with neat diagram Peritectic transformations, Eutectic transformation and Eutectoid transformations	Onder stand	CAMEUUS.U/	
9	Compare annealing and normalizing, annealing and hardening.	Remember	CAME005.07	
10	Explain following with neat diagrams Annealing, Normalising,	Understand		
	Process annealing, Stress relieving and hardening.	2 11301 Stuffe	311111111111111111111111111111111111111	
	UNIT-IV			
	ENGINEERING MATERIALS-II,III			
	Part – A (Short Answer Questions)			
1	What is cast iron? Classify the cast iron.	Understand	CAME005.08	
2	Explain about is gray cat iron?		CAME005.08	

3	Define white cast iron. Explain its properties?	Understand	CAME005.08
4	Explain about is Malleable cast iron?	Remember	
5	What is spheroidal cast iron?	Understand	
6	Examine carbon % present in malleable cast iron?	Understand	
7	Examine carbon % present in maneagre cast from?	Remember	CAME005.08
8	Examine carbon % present in Spherotata cast from:	Understand	
9	Discuss about microstructure of gray cast iron?	Remember	CAME005.08
10	Draw the microstructure of white cast iron and explain.	Understand	
11	Explain about microstructure of spheroidal cast iron?	Understand	
12	Where is titanium alloys used?	Remember	CAME005.08
13	Explain about microstructure of malleable cast iron?	Understand	
14	What are the applications of aluminium alloys?	Remember	CAME005.16
15	Explain about are heat treatable alloys?	Remember	CAME005.08
16	Explain about are non-heat treatable alloys?	Understand	
17	Classify aluminium alloys and explain their properties in detail.	Understand	
18	Explain about are the important copper alloys?	Understand	
19	Classify Titanium alloys with their properties and applications.	Remember	CAME005.08
20	Explain about the applications of copper alloys?	Remember	CAME005.15
20	Part – B (Long Answer Questions)	Remember	CAMEOUS.13
1	Explain types of cast irons and give applications for each.	Understand	CAME005.08
2	Describe methods of making white cast iron? Explain properties.	Understand	
3	Justify "ti-6al-4v alloys are useful for aerospace applications"	Understand	
4	Explain about gray cast iron and explain properties?	Understand	
5	Discuss the advantages of steels over the family of cast iron.	Remember	CAME005.08
6	Why nodular cast iron is ductile. Explain properties?	Understand	
7	Discuss effect of Silicon, Manganese, Sulphur, and Phosphorous in	Understand	
,	cast iron.	Onderstand	C/ HVIL003.00
8	Discuss Ni-resist and Ni-hard cast irons.	Remember	CAME005.08
9	Explain heat treatable aluminium alloys and give applications.	Understand	
10	Discuss non-heat treatable aluminium alloys and give applications.	Remember	CAME005.16
11	What are the different copper alloys and give applications for each.	Understand	CAME005.16
12	Explain about are types of brasses and explain properties?	Understand	CAME005.16
13	Discuss precipitation hardening with an example.	Understand	CAME005.08
14	Explain various bronzes and their properties.	Understand	CAME005.08
15	Explain beta titanium alloys.	Understand	CAME005.08
16	Classify titanium alloys and give typical applications for each.	Remember	CAME005.14
17	Explain about soldering alloys? Give typical applications.	Understand	CAME005.14
18	Write the properties, applications of alpha titanium alloys.	Remember	CAME005.13
19	Explain properties and applications of duraluminum.	Remember	CAME005.15
20	Compare naval brass and muntz metal.	Understand	CAME005.08

Part – C (Problem Solving and Critical Thinking)			
1	Compare gray and white cast iron.	Understand	CAME005.08
2	Explain about factors control structure of cast iron?	Understand	CAME005.08
3	Compare malleable and nodular cast iron.	Understand	CAME005.08
4	Explain about dominant mechanical property of cast iron?	Remember	CAME005.08
5	Compare brasses and bronzes.	Understand	CAME005.08
6	Describe the following cast irons White cast iron, Malleable cast iron,	Understand	CAME005.08
	Gray cast iron and Nodular cast iron		
7	Explain about effects of adding Si, Mn, S, and P?	Remember	CAME005.08
8	Compare heat treatable and non-heat treatable aluminium alloys.	Understand	CAME005.08
9	Explain about effects of graphite flakes? Explain in detail about	Understand	CAME005.08
	ductility nature of nodular cast iron.		
10	Compare pure titanium and Ti-6Al-4 V.	Remember	CAME005.08
	UNIT-V		
	ENGINEERING MATERIALS-IV		
~	Part - A (Short Answer Questions)		I ~
S No	QUESTION	Blooms Taxonomy	Course Learning
		level	Outcomes
1	Classify ceramics. Give the examples.	Understand	
2	Explain about uses of alumina ceramics?	Understand	CAME005.10
3	Explain about uses of Zirconium ceramics?	Understand	CAME005.10
4	Explain about uses of Silicon nitride?	Remember	CAME005.10
5	Classify Glasses and give their properties and applications.	Understand	CAME005.10
6	Define glass. Explain properties of glass.	Understand	CAME005.10
7	Explain about glass transition temperature?	Remember	CAME005.10
8	Explain about borosilicate glass?	Understand	CAME005.10
9	Compare soft and hard glasses.	Remember	CAME005.10
10	Explain about thermal shock?	Understand	CAME005.10
11	Classify polymers. Give the examples.	Understand	CAME005.09
12	Explain about polymerisation?	Understand	CAME005.09
13	What is a thermo plastic? Explain?	Understand	CAME005.09
14	Explain about thermo setting plastics?	Remember	CAME005.09
15	What are the different types of additives used in polymers?	Understand	CAME005.09
16	What is FRP? Explain?	Understand	CAME005.09
17		D 1	CAME005.11
	Define composites and advantages of composites.	Remember	CAME003.11
18	Define composites and advantages of composites. Classify composites with examples.	Understand	
18 19			
	Classify composites with examples.	Understand	CAME005.11 CAME005.11
19	Classify composites with examples. Explain about matrix in composite?	Understand Remember	CAME005.11 CAME005.11

2	Classify ceramics with examples and applications.	Understand	CAME005.14
3	Explain about properties and applications of alumina ceramics.	Understand	CAME005.15
4	Discuss about properties and applications of zirconia ceramics.	Remember	CAME005.13
5	Explain about properties and applications of silicon carbide.	Understand	CAME005.14
6	Discuss about properties and applications of silicon nitride.	Understand	CAME005.15
7	Explain about properties and applications of tungsten carbide.	Remember	CAME005.16
8	Define glass and classify types of glasses and give applications.	Understand	CAME005.16
9	Explain glass transition temperature and its importance in	Remember	CAME005.10
	manufacture.		
10	Compare properties of hard and soft glasses.		CAME005.10
11	Explain about is stabilization in zirconia ceramics?	Remember	
12	Explain brittleness in ceramic materials.		CAME005.10
13	Define polymers. Classify them and give typical applications.	Understand	CAME005.09
14	Discuss methods of polymerisation.	Remember	CAME005.08
15	Explain about are thermo plastics? Explain properties.		CAME005.13
16	Explain about are thermo setting plastics? Give their properties.	Understand	CAME005.13
17	Classify composites. Explain their properties.	Remember	CAME005.09
18	Discuss fibre reinforced plastics and their uses.	Understand	CAME005.16
19	Give properties of metal matrix composites.	Understand	CAME005.16
20	Explain ceramic matrix composites and give their properties.	Understand	CAME005.16
	Part – C (Problem Solving and Critical Thinking)		
1	Explain alumina ceramics and glasses.	Understand	CAME005.10
2	Compare CBN and silicon nitride.	Remember	CAME005.10
3	Differentiate Zirconia and alumina ceramics.	Understand	CAME005.10
4	Compare hard and soft glasses.	Remember	CAME005.10
5	Explain about various factors for thermal shock resistance?	Understand	CAME005.10
6	Differentiate metal matrix and ceramic matrix composites.	Understand	CAME005.10
7	Compare composites and polymers.	Remember	CAME005.16
8	Explain and differentiate FRP and PVC.	Understand	CAME005.09
9	Compare polymer matrix and ceramic matrix composites.	Remember	CAME005.09
10	Differentiate tungsten carbide and silicon nitride.	Understand	CAME005.10

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